#### **SEMESTER-I**

# ANALYTICAL TECH& SPECTROSCOPY-I PRACTICAL SYLLABUS

Course Code:P20/CHE/DSC/104/P Max Marks :50

Course Type: DSC-4

No. of Hrs/Week: 4 Hrs

No. of Credits:2

## **COURSE OUTCOME:**

**CO1:** Calibration of weights, pipettes, standard flasks, burette

CO2: The student should have knowledge on Data analysis, Significant figures, Precision and accuracy

**CO3:** The student will be able to understand practical knowledge on potentiometric titrations with respect to strong acid and strong base.

**CO4:** Determination of specific rotation of sucrose

#### 1. Calibrations:

- (i) Calibration of weights.
- (ii) Calibration of pipettes.
- (iii) Calibration of standard flasks.
- (iv) Calibration of burette.

## **Physical properties:**

Data analysis I: Significant figures, Precision and accuracy

## **Potentiometry:**

Titration of strong acid vs strong base
Titration of weak acid vs strong base
Determination of dissociation constant of a weak acid
Determination of single electrode potential

## **Polarimetry:**

Determination of specific rotation of sucrose Acid-catalyzed hydrolysis of sucrose (inversion of sucrose)

# **Reference Books:**

- 1. Khosla, B.D., Garg, V.C., and Khosla, A. Senior Practical Physical Chemistry.
- 2. Athawale, V., and Mathur, P. Experimental Physical Chemistry.
- 3. Vishwanathan, B., and Raghavan, P.S. Practical Physical Chemistry.
- 4. Sindhu, P.S. Practical in Physical Chemistry.
- 5. Yadav, J.B. Advanced Practical Physical chemistry.
- 6. Vogel. (2002). Text book of Quantitative Analysis (6th ed). Pearson educationLtd.

#### **SEMESTER-I**

## **ANALYTICAL TECH & SPECTROSCOPY - I**

# MODEL PRACTICAL QUESTION PAPER

Course Code:P20/CHE/DSC/104/P Max Time: 3Hrs
Credits: 2 Max Marks: 50 Marks

1) Write the Principle involved in the given Potentiometric/Polarimetric Experiments.(**CO4**)

10 M

2) a) Determine the strength of the given acid using Potentiometer.(CO3)

OR

b) Determine the specific rotation of Optically active Compounds(CO4).

OR

c)Inversion of cane sugar by Polarimetry. (CO3 & CO4)

25 M

3) Record and Attendance

5 M

4) Viva Voce (**CO1,CO2, CO3 & CO4**)

10 M